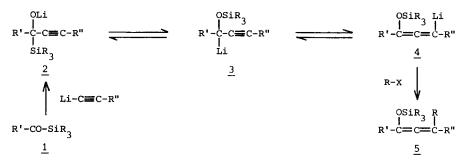
ANIONIC REARRANGEMENTS STARTING WITH 1-LITHIO-2-PROPYNYL SILYL ETHERS A ROUTE TO (β-TRIMETHYLSILYLVINYL) SILYL KETONES

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Abstract The title compounds undergo silyl-Wittig rearrangement at low temperature and lead to allenolates at room temperature.

The mobility of both silicon and lithium in the system 2 = 3 = 4 forms the basis of syntheses of siloxyallenes 5 (and the corresponding α,β -unsaturated ketones) from silyl ketones $\underline{1}^1$.



We have studied three cases in which R'= H and have found that

- (a) this system can be entered at <u>3</u> through lithiation of 2-propynyl trialkyl silyl ethers
 (<u>6</u> a-c);
- (b) yet another mode of mobility in propargylic-allenic lithium compounds 1,3-migration of hydrogen² - leads to a fourth type of tautomer, the allenolates <u>7</u> a-c.

The transformation $6 \rightarrow 7$ is effected most advantageously in the case of <u>6a</u>. Under the same circumstances (*vide infra*) slower deprotonation of <u>6b</u> and faster silyl ether cleavage of <u>6c</u> cause lower yields (15% and 30%, respectively) of the final products 8b and 8c derived from <u>7b</u> and <u>7c</u> by hydrolysis.

Treatment of <u>6a</u> with *n*-BuLi at -30° C in ether-hexane (3:1) led to <u>2a</u>. Standing at room temperature overnight converted this alkoxide into the allenolate <u>7a</u>. Hydrolysis of <u>7a</u> yielded a 85:15 mixture of the Z/E isomers of <u>8a</u> (73%, based on <u>6a</u>)³. The isomers can be separated by column chromatography on silica. The main product <u>Z-8a</u> isomerizes to <u>E-8a</u> on treatment with acid or on standing in CDCl₂.

<u>7a</u> is attacked at oxygen by alkyl halides and by trimethylchlorosilane, respectively. The resulting allenol ethers can be hydrolyzed with acid to give $E-\underline{8a}$. The mode of formation of <u>7</u> has not yet been established. However, a likely route involves reversion of <u>2</u> (R' = H) to its type-3 or type-4 isomer, which can undergo 1,3-migration of hydrogen² to give <u>9</u>. Silyl-Wittig rearrangement of <u>9</u> gives <u>7</u>.Lithium 3-trimethylsilyl-2-propynoxide did not undergo 1,3-migration of hydrogen when subjected to the conditions of formation of 7a.

<u>Sa</u> and <u>Sc</u> have the interesting feature of being both an acylsilane and a vinylsilane and thus can be expected to be prone to selective modification at either carbon-silicon moiety by nucleophilic or electrophilic agents, respectively⁴.

References

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 H.J. Reich and M.J. Kelly, J. Am. Chem. Soc., <u>104</u>, 1119 (1982).
- See J. Klein, in The Chemistry of the Carbon-Carbon Triple Bond, S. Patai, Ed., John Wiley and Sons, 1978, part 1, p. 343.
 Recent example: P.E. van Rijn, R.H. Everhardus, and L. Brandsma, Recl. Trav. Chim. Pays-Bas, <u>99</u>, 179 (1980).
- 3. $Z-\underline{8a}$: $IR (CCl_4, cm^{-1})$: 1620, 1540; ¹H-NMR (CDCl_3, δ): 7,32 (d, J=13,5 Hz, 1H), 6,05 (d, J=13,5 Hz, 1H), 0,94 (s, 9H), 0,22 (s, 6H), 0,14 (s, 9H); ¹³C-NMR (CDCl_3, δ): 237,9 (s), 144,8 (d), 144,4 (d), 26,4 (q), 16,8 (s), -0,7 (q), -7,1 (q); MS (70 eV, m/e): 227 (0.1%), 186 (1%), 73 (100%).

 $E-\underline{8a}: IR (CCl_4, cm^{-1}): 1620, 1590, 995; {}^{1}H-NMR (CDCl_3, \delta): 6,86 (d, J=19,6 Hz, 1H), 6,72 (d, J=19,6 Hz, 1H), 0,93 (s, 9H), 0,24 (s, 6H), 0,14 (s, 6H); {}^{13}C-NMR (CDCl_3, \delta): 236,3 (s), 147,0 (d), 145 (d), 26,6 (q), 16,6 (s), -1,8 (q), -5,9 (q). MS (70 eV, m/e): 227 (0,2 %), 186 (3%), 73 (100%).$

In the cases of $\underline{8b}$ and $\underline{8c}$ the Z/E-ratios were 90:10. Analytical data of these compounds were in complete agreement with the structures assigned to them.

 For nucleophilic attack on acylsilanes cf. ref. 1 and: D. Schinzer and C.H. Heathcock, Tetrahedron Letters, 1881 (1981).

For electrophilic attack on vinylsilanes cf. T.H. Chan and I. Fleming, Synthesis, 761 (1979).

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